

Synthesis and characterization of some new 2(3H)-benzoxazolones with analgesic and antiinflammatory activities

NESRİN GÖKHAN-KELEKÇİ¹, MERİÇ KÖKSAL², SONGÜL ÜNÜVAR³,
GÖKNUR AKTAY³, & HAKKI ERDOĞAN¹

¹Department of Pharmaceutical Chemistry, Faculty of Pharmacy, Hacettepe University, 06100 Sıhhiye, Ankara, Turkey,

²Faculty of Pharmacy, Yeditepe University, 34755 Kayışdağı, Istanbul, Turkey, and ³Department of Pharmacology, Faculty of Pharmacy, İnönü University, 44280 Malatya, Turkey

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Abstract

The synthesis, characterization and pharmacological activities of a new series of (6-difluorobenzoyl)-5-methyl-3-benzoylmethyl-2(3H)-benzoxazolone and 5-methyl-3-(2-hydroxyl-2-phenylethyl)-2(3H)-benzoxazolone are described. Antiinflammatory activity was investigated by the carrageenin-induced paw oedema test and analgesic activity by acetic acid writhing and hot plate tests in mice. Among the synthesized compounds, compound **3e** 6-(2,5-difluorobenzoyl)-3-(4-bromobenzoylmethyl)-2(3H)-benzoxazolone was found to be the most promising compound for analgesic activity. Reduced compounds (**4a–4d**) displayed considerable anti-inflammatory activity compared to the other derivatives.

Keywords: 2(3H)-Benzoxazolone, analgesic-antiinflammatory activities

Introduction

Non-steroidal anti-inflammatory drugs (NSAIDs) are a mainstay in the treatment of inflammation and they owe their therapeutic and side effects in large part to the inhibition of cyclooxygenase (COX). The separation of the therapeutic effects from the side effects has been a major challenge in the design and synthesis of these drugs. The discovery of a second isoform of cyclooxygenase, namely COX-2, has opened a new line of research based on the assumption that pathological prostaglandins are produced by the inducible isoform COX-2 while physiological prostaglandins are produced by the constitutive isoform COX-1 [1]. On this premise several new inhibitors have been developed and some are now commercially available [2–3]. However, an increased risk of myocardial infarction and cardiovascular thrombotic events associated with the use of some selective COX-2 inhibitors has been observed [4]. These adverse

cardiovascular effects, which are attributed to a decreased level of the vasodilatory PGI₂ and an increased level of the potent platelet aggregator TxA₂, were primarily responsible for the recent withdrawal of rofecoxib (Vioxx[®]) and valdecoxib (Bextra[®]) from the market [5]. On our going medicinal chemistry research programme, we have seen that 2(3H)-benzoxazolones may be a building block in the drug structures and these structures bearing 2(3H)-benzoxazolones exhibit wide range of biological activities such as analgesics-antiinflammatory [6–16], dopamine receptor agonist [17], cardiostonic [18], antihypertensive [19], and antiulcer [20]. In our laboratory, we have designed and synthesized some 2-benzoxazolinone derivatives in the search for new non-steroidal anti-inflammatory agents [6–9,12,14]. A considerable number of the prepared compounds have been found to have analgesic-anti-inflammatory activity comparable to or higher than that of indomethacin. In this paper,

Correspondence: N. Gökhan-Kelekçi, Department of Pharmaceutical Chemistry, Faculty of Pharmacy, Hacettepe University, 06100 Sıhhiye, Ankara, Turkey. E-mail: onesrin@hacettepe.edu.tr

we report the synthesis and pharmacological screening of a new series of 2(3*H*)-benzoxazolones in order to go deep into structure-activity relationships for further studies. On this basis we have synthesized some 5-methyl-3-benzoylmethyl-2(3*H*)-benzoxazolones, reduced derivatives of them 5-methyl-3-(2-hydroxyl-2-phenylethyl)-2(3*H*)-benzoxazolones and 6-(difluorobenzoyl)-3-benzoylmethyl-2(3*H*)-benzoxazolones. The synthesized compounds were tested for their analgesic and antiinflammatory activities.

Experimental

Chemistry

Melting points were determined through a Thomas Hoover capillary melting point apparatus and are uncorrected. Infrared (IR) spectra were obtained with a Bruker Vector 22 IR (Opus Spectroscopic Software Version 2.0) spectrometer by using potassium bromide plates and expressed in wave number (cm^{-1}). Nuclear magnetic resonance ($^1\text{H-NMR}$ and $^{13}\text{C-NMR}$) spectra were scanned on a Bruker 80 MHz spectrometer using chloroform as solvent. Chemical shifts are expressed in δ (parts per million) relative to tetramethylsilane. The mass spectra were obtained with electron impact technique by a Direct Insertion Probe and Agilent 5973-Network Mass Selective Dedector at 70 eV. Elemental analyses (C, H, N) were performed on Leco CHNS 932 analyzer.

Synthesis of 5-methyl-2(3H)-benzoxazolone. 5-Methyl-2(3*H*)-benzoxazolone was synthesized as a result of the reaction of 5-methyl-2-hydroxyaniline with urea in oil bath according to the method reported earlier [21].

5-methyl-3-benzoylmethyl-2(3H)-benzoxazolone (Compound 3a–3d). To the solution of sodium 5-methyl-2(3*H*)-benzoxazolones in alcohol, equimolar α -bromo-4-substituted acetophenone was added. The solution was refluxed on an oil bath for 4 h and then cooled. The crude product was filtered, washed with water, dried and crystallized from a suitable solvent.

6-Difluorobenzoyl-2(3H)-benzoxazolone. 6-Difluorobenzoyl-2(3*H*)-benzoxazolone was synthesized by treating 2(3*H*)-benzoxazolone and difluorobenzoic acid in polyphosphoric acid by the method reported earlier [22].

5-methyl-3-(2-hydroxyl-2-phenylethyl)-2(3H)-benzoxazolone (Compound 3e–3h, 4a–4d). 0.01 mol 5-methyl-3-benzoylmethyl-2(3*H*)-benzoxazolone was dissolved in methanol. To this mixture 0.012 ml

of NaBH_4 was added, and stirring was continued for 4 h at reduced pressure. The residue was recrystallized from different solvents.

5-methyl-3-benzoylmethyl-2(3H)-benzoxazolone (3a). Yield: 32%, mp 125°C. IR (KBr): ν 3061, 2967, 2927, 1774, 1689. $^1\text{H-NMR}$ (80 MHz, CDCl_3): δ 2.2 (3H, s, CH_3), 5.1 (2H, s, N-CH_2), 6.5–8.0 (8H, m, Arom.H). EI (70 eV): m/z : 267 (M^+), 162, 105, 77. Analysis calculated for $\text{C}_{16}\text{H}_{13}\text{NO}_3$ (267.28) Anal. found: C, 71.7; H, 4.98; N, 5.51; Anal. calcd: C, 71.9; H, 4.90; N, 5.24%.

5-methyl-3-(4-methylbenzoylmethyl)-2(3H)-benzoxazolone (3b). Yield: 40%, mp 159°C. IR (KBr): ν 3054, 2969, 2921, 1763. $^1\text{H-NMR}$ (80 MHz, CDCl_3): δ 2.2 (3H, s, CH_3 -phenyl), 2.3 (3H, s, CH_3); 5.1 (2H, s, N-CH_2); 6.5–7.9 (7H, m, Arom.H). EI (70 eV): m/z : 281 (M^+), 119, 91. Analysis calculated for $\text{C}_{17}\text{H}_{15}\text{NO}_3$ (281.31) Anal. found: C, 72.69; H, 4.85; N, 4.94; Anal. calcd: C, 72.58; H, 5.37; N, 4.98%.

5-methyl-3-(4-bromobenzoylmethyl)-2(3H)-benzoxazolone (3c). Yield: 40%, mp 171°C. IR (KBr): ν 2963, 2931, 1759, 1696. $^1\text{H-NMR}$ (80 MHz, CDCl_3): δ 2.3 (3H, s, CH_3), 5.1 (2H, s, N-CH_2), 6.6–8.0 (7H, m, Arom.H). EI (70 eV): m/z : 345 (M^+), 347 ($\text{M} + 2$), 183, 185 (base peak), 162, 91. Analysis calculated for $\text{C}_{16}\text{H}_{12}\text{BrNO}_3$ (346.18) Anal. found: C, 63.76; H, 3.46; N, 4.63; Anal. calcd: C, 63.69; H, 4.01; N, 4.64%.

5-methyl-3-(4-chlorobenzoylmethyl)-2(3H)-benzoxazolone (3d). Yield: 56%, mp 164°C. IR (KBr): ν 3040, 2964, 2928, 1758, 1694. $^1\text{H-NMR}$ (80 MHz, CDCl_3): δ 2.3 (3H, s, CH_3), 5.1 (2H, s, N-CH_2), 6.6–8.05 (7H, m, Arom.H). EI (70 eV): m/z : 301 (M^+), 303 ($\text{M} + 2$), 162, 139, 141, 111, 113, 91. Analysis calculated for $\text{C}_{16}\text{H}_{12}\text{ClNO}_3$ (301.73) Anal. found: C, 63.76; H, 3.46; N, 4.63; Anal. calcd: C, 63.69; H, 4.01; N, 4.64%.

6-(2,5-Difluorobenzoyl)-3-(4-bromobenzoylmethyl)-2(3H)-benzoxazolone (3e). Yield: 85%, mp 217°C, Ref [22].

6-(2,5-Difluorobenzoyl)-3-(4-chlorobenzoylmethyl)-2(3H)-benzoxazolone (3f). Yield: 74%, mp 227°C, Ref [22].

6-(2,6-Difluorobenzoyl)-3-(4-bromobenzoylmethyl)-2(3H)-benzoxazolone (3g). Yield: 67%, mp 240°C, Ref [22].

6-(2,6-Difluorobenzoyl)-3-(4-chlorobenzoylmethyl)-2(3H)-benzoxazolone (3e). Yield: 77%, mp 217°C, Ref [22].

5-methyl-3-(2-hydroxyl-2-phenylethyl)-2(3H)-benzoxazolone (4a). Yield: 99%, mp 103–4°C. IR (KBr): ν 3467, 1750. $^1\text{H-NMR}$ (80 MHz, CDCl_3): δ 2.3 (3H, s, CH_3); 3.95 (2H, d, N-CH_2); 5.05 (1H, t,

CH₂—CH), 6.8–7.6 (8H, m, Arom.H). EI (70 eV): m/z: 163, 134, 121, 91, 77. Analysis calculated for C₁₆H₁₅NO₃ (269.30) Anal. found: C, 71.48; H, 5.96; N, 4.92; Anal. calcd: C, 71.36; H, 5.61; N, 5.20%.

5-methyl-3-[2-hydroxyl-2-(4-methylphenyl)ethyl]-2(3H)-benzoxazolone (4b). Yield: 92%, mp 105–6°C. IR (KBr): ν 3450, 1750. ¹H-NMR (80 MHz, CDCl₃): δ 2.3 (6H, s, CH₃, CH₃-fenil); 4.0 (2H, d, N—CH₂—); 5.05 (1H, t, CH₂—CH), 6.7–7.5 (7H, m, Arom.H). EI (70 eV): m/z: 163, 134, 121, 91, 77. Analysis calculated for C₁₇H₁₇NO₃ (283.33) Anal. found: C, 72.03; H, 5.46; N, 4.93; Anal. calcd: C, 72.07; H, 6.05; N, 4.94%.

5-methyl-3-[2-hydroxyl-2-(4-bromophenyl)ethyl]-2(3H)-benzoxazolone (4c). Yield: 99%, mp 148–9°C. IR (KBr): ν 3430, 1745. ¹H-NMR (80 MHz, CDCl₃): δ 2.3 (3H, s, CH₃); 4.0 (2H, d, N—CH₂—); 5.1 (1H, t, CH₂—CH), 6.7–7.6 (7H, m, Arom.H). EI (70 eV): m/z: 347 (M⁺), 348 (M + 2), 185, 187, 163, 134, 91, 77. Analysis calculated for C₁₆H₁₄BrNO₃ (348.20). Anal. found: C, 55.46; H, 4.26; N, 4.32; Anal. calcd: C, 55.19; H, 4.05; N, 4.02%.

5-methyl-3-[2-hydroxyl-2-(4-chlorophenyl)ethyl]-2(3H)-benzoxazolone (4d). Yield: 97%, mp 139–140°C. IR (KBr): ν 3463, 1742. ¹H-NMR (80 MHz, CDCl₃): δ 2.3 (3H, s, CH₃); 4.0 (2H, d, N—CH₂—); 5.1 (1H, t, CH₂—CH), 6.7–7.5 (7H, m, Arom.H). EI (70 eV): m/z: 303 (M⁺), 305 (M + 2), 163, 134, 91, 77. Analysis calculated for C₁₆H₁₄ClNO₃ (303.75). Anal. found: C, 63.58; H, 4.99; N, 4.63; Anal. calcd: C, 63.27; H, 4.65; N, 4.61%.

Pharmacology

Animals. Male Swiss albino mice (20–25 g) were purchased from the animal breeding laboratories of Firat University (Elazig, Turkey). Before treatments, the animals were acclimatized to animal laboratory conditions for two days and were fed on standard pellet diet and water ad libitum. On the day before the treatments, the food was withdrawn, but the animals were allowed free access of water. A minimum of five animals was used in each group, otherwise it is described in the procedure. Mice used in the present study were cared for in accordance with the directory of Firat University Animal Care Unit, which applies the guidelines of National Institutes of Health on laboratory animal welfare.

Preparation of test samples for bioassay. Test samples were given orally to test animals after suspending in 0.5% sodium carboxymethyl cellulose (CMC) and distilled water. The control group animals received the same experimental handling as those of the test groups except that the drug treatment was replaced with appropriate

volumes of the dosing vehicle. Either indomethacin (10 mg/kg), or acetylsalicylic acid (ASA) (200 mg/kg) in 0.5% CMC was used as reference drug.

Anti-inflammatory activity

Carrageenan induced oedema [23]. For the determination of the effects on carrageenan-induced paw oedema the modified method of Kasahara et al. was employed [23]. One hour after the oral administration of either test sample or dosing vehicle, each mice was injected with freshly prepared (0.5 mg/25 μ L) suspension of carrageenan (Sigma, St. Louis, Missouri, U.S.A.) in physiological saline (154 mM NaCl) into subplantar tissue of the right hind paw. As the control, 25 μ L saline solution was injected into that of the left hind paw. Paw oedema was measured in every 90 min during 6 h after induction of inflammation. The difference in footpad thickness between the right and left foot was measured with a pair of dial thickness gauge callipers (Ozaki Co., Tokyo, Japan). Mean values of treated groups were compared with mean values of a control group and analyzed using statistical methods. Indomethacin was used as a reference compound and administered at 10 mg/kg.

Analgesic activity

Koster test [24]. One hour after the oral administration of test sample, each mouse was injected with 3% (w/v) acetic acid solution (0.1 mL/10 g body weight) intraperitoneally. Starting 5 min after the acetic acid injection, the number of muscular contractions on mice were counted for a period of 10 min. A significant reduction in the number of writhings by any treatment as compared to control animals was considered as a positive analgesic response. The antinociceptive activity was expressed as percentage change from writhing controls. Percent inhibitory effects were estimated according to the following equation, where n was the average difference in thickness between the left and right hind paw of control group and n' was that of test group of animals. Aspirin (ASA) was used as a reference compound and administered at 200 mg/kg.

$$\text{Inhibition(\%)} = [(n - n')/n] \times 100$$

Constant temperature hot-plate test [25]. An HTC Inc. Mod. 35-D analgesiameter was set to give a plate temperature of $54 \pm 0.5^\circ\text{C}$. One hour after oral administration of the compounds (100 mg/kg), the animals were placed on the hot-plate and confined by a lidded perspex box in a compartment measuring 13.8×13.8 cm, and the latency to the first hind-paw lick was recorded. If no hind-paw lick occurred, the

test was terminated after 30 s. Morphine was used as a reference analgesic and administered i.p. at 10 mg/kg.

Statistical analysis of data. Data obtained from animal experiments were expressed as means \pm standard error (SEM). Differences between the control and treated groups were tested for significance using a one-way analysis of variance (ANOVA), followed by Dunnett's *t*-test.

Result and discussion

Chemistry

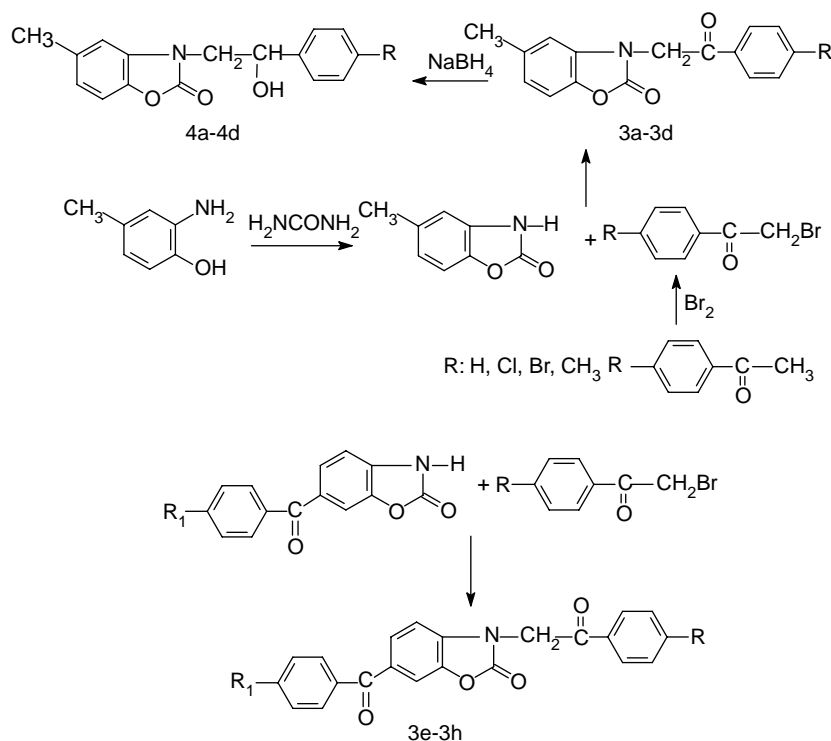
5-Methyl-3-(4-substituted benzoylmethyl)-2(3*H*)-benzoxazolones (**3a–3d**) were prepared by reacting 5-methyl-2(3*H*)-benzoxazolones, α -bromo-4-substituted acetophenones and sodium hydroxide in ethanol. Compounds **4a–4d** were obtained by the reduction of ketones (**3a–3d**) with NaBH₄. Compounds **4a–4d** have one asymmetric carbon atom. However the enantiomers were not separated. Studies directed toward separation of enantiomers will be done in the near future. 6-Acyl-2(3*H*)-benzoxazolones were prepared by reacting 2(3*H*)-benzoxazolone with difluorobenzoic acids in the presence of polyphosphoric

acid at 140–160°C. The other compounds **3e–3h** were prepared from 6-acyl-2(3*H*)-benzoxazolone and α -bromo-4-substituted acetophenone [22]. The synthetic pathway for preparation of the targeted compounds is shown in Scheme 1.

The structure of the compounds was confirmed by IR, ¹H-NMR, Mass and elemental analysis. In IR spectra of the compounds, the bands seen at 1742–1774 cm⁻¹ (lactam C=O) and 1682–1696 cm⁻¹ (CH₂-C=O) were in accordance with the assumed structures. Furthermore, N–H stretching bands belonging to 2(3*H*)-benzoxazolone ring disappeared with the reaction of benzoylmethyl bromides. The formation of compounds **4a–4d** were confirmed by the presence of OH signals at 3030–3070 cm⁻¹ in the IR spectra.

In the ¹H-NMR spectra of the compounds, 5-methyl protons bound to 2(3*H*)-benzoxazolone appeared at approximately 2.2–2.3 ppm. The methylene protons seen as singlet at 5.1 ppm proved the presence of benzoylmethyl moiety. The same methylene protons were observed in 3.95–4.0 ppm in reduced derivatives **4a–4d**. The triplets of (–CH) methyne protons of the compounds were found to be 5.05–5.1 ppm.

The mass spectra of the compounds were studied under electron ionization. Molecular ion M⁺ peaks which appeared at different intensities confirmed the molecular weights of the examined compounds except



3a: R= H, **3b:** R= CH₃, **3c:** R= Br, **3d:** R= Cl, **3e:** R= Br; R₁= 2,5-diF,
3f: R= Cl; R₁= 2,5-diF, **3g:** R= Br; R₁= 2,6-diF, **3h:** R= Cl; R₁= 2,6-diF
4a: R= H, **4b:** R= CH₃, **4c:** R= Br, **4d:** R= Cl

Scheme 1. Protocol for synthesis of titled compounds **3a–3h**, **4a–4d**.

compound **4a**. Characteristic $M + 2$ isotop peaks are observed in the mass spectra of the compounds having a halogen. In the spectra of compounds **3a–3d**, the base peak resulted from loss of N,5-dimethyl-2(3H)-benzoxazolone while in the spectra of compound **4a–4d** the lost group was the base peak (Scheme 2). In elemental analysis the results support the structures with $\pm 0.4\%$ of the theoretical values.

Pharmacology

Antiinflammatory activity. In order to screen the anti-inflammatory profile of the synthesized compounds, carrageenan-induced hind paw oedema model in mice was used [23]. Since the carrageenan oedema has been used in the development of indomethacin, many researchers have adapted this procedure for screening potential anti-inflammatory compounds. Carrageenan-induced oedema is a nonspecific inflammation maintained by the release of histamine, 5-hydroxytryptamine, kinins and later by prostaglandins [26]. The inhibitory effect of NSAIDs, such as indomethacin, is usually weak in the first phase (1–2 h), in contrast with their strong inhibition in the second phase (3–4 h) [27].

The anti-inflammatory activity of the synthesized compounds was studied at 100 mg/kg dose. Test compounds which possessed more than 20% effect, even some of them are not be significant statistically, were considered for further evaluation and the experiments were repeated for the compounds in two different dose levels (50 and 200 mg/kg). A notable

decrease in activity was seen compounds **3a–3d** bearing 4-substituted benzoylmethyl in the third position of 2(3H)-benzoxazolone. When compared the ketone and alcohol groups, It was seen that reduced derivatives (**4a–4d**) were more effective than those of ketone (**3a–3d**) (Table I, Figure 1–3).

Good inhibition of the second phase of carrageenan-induced oedema was observed for the compounds **4a, 4c, 4d** tested, suggesting that they interfere with prostaglandin synthesis. 90 and 180 minutes after the administration of the drug, compound **4a** bearing no substituent at the phenyl ring exhibited much significant activity at 50 and 100 mg/kg dose when compared to indomethacin. However, 360 minutes after the administration of the drug, the same compound **4a** showed half activity of that of indomethacin at 100 mg/dose. It was shown that compound **4c** carrying brom substituent at 100 mg/kg dose and compound **4d** bearing chloro substituent at 200 mg/kg dose have more significant activity among the reduced derivatives.

Analysis of the activities of the compounds **3e–3h** bearing substituted benzoyl methyl in 3 position and difluorobenzoyl in 6 position of the 2(3H)-benzoxazolone moiety showed the difference in the anti-inflammatory action due to the substituents in the benzoylmethyl group and position of fluors in the benzoyl group. As a result of this, the compounds possessing 4-bromobenzoylmethyl and 2,6-difluorobenzoyl moieties were more active than the others including 2,5-difluorobenzoyl and 4-chlorobenzoylmethyl (Table I).

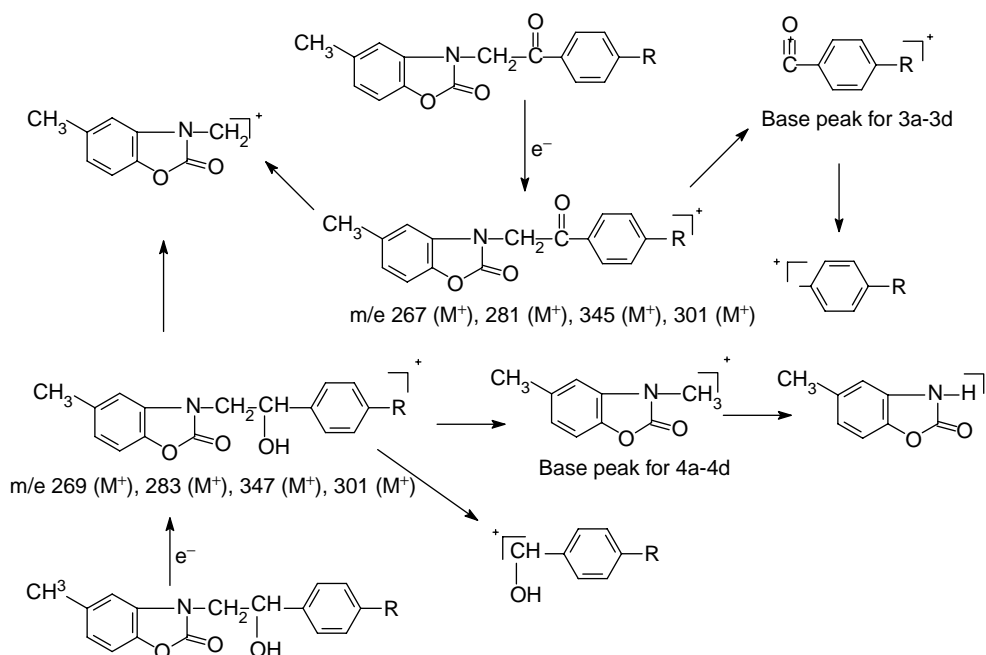


Table I. Analgesic/anti-inflammatory activities for compounds **3a–3h**, **4a–4d**.

Compounds	R	Dose mg/kg	Abdominal constriction \pm SEM (Inhibition %)	Hind paw lick \pm SEM (Inhibition %)	Dose mg/kg	Swelling in thickness ($\times 10^{-2}$ mm) \pm SEM (Inhibition %)			
						90 min.	180 min.	270 min.	360 min.
3a	H	100			100	50.0 \pm 5.0 (6.9)	101.6 \pm 13.0 (–17.8)	8.3 \pm 10.4** (–35.5)	73.3 \pm 7.2 (–6.6)
3b	CH ₃	100			100	61.6 \pm 7.2 (–14.7)	80 \pm 7.6 (7.1)	101.6 \pm 14.8 (–40.1)	88.3 \pm 13.0 (–28.0)
3c	Br	100			100	57.5 \pm 7.5 (–7.0)	93.3 \pm 7.2 (–8.2)	113.1 \pm 10.1* (–56.2)	83.3 \pm 4.4 (–21.2)
3d	Cl	100			100	53.3 \pm 1.6 (0.7)	93.3 \pm 3.3 (–8.2)	125 \pm 8.6** (–72.4)	98.3 \pm 13.6 (–43)
3e	Lit	100	31.7 \pm 5.3** (41.7)	5.8 \pm 0.3** (29.3)	100	41.0 \pm 5.5 (23.6)	101.4 \pm 3.9 (–17.6)	90.3 \pm 3.4 (–24.6)	80.5 \pm 1.3 (–17.3)
					50	55.3 \pm 1.2 (–2.9)	92.3 \pm 12.9 (–7.0)	78.7 \pm 6.7 (–8.6)	75.7 \pm 5.8 (–10.3)
					200	45.5 \pm 4.3 (15.3)	91.5 \pm 3.1 (–6.1)	72.8 \pm 4.4 (–0.4)	86.6 \pm 7.1 (–26.1)
3f	Lit	100			100	49.5 \pm 2 (7.8)	87.7 \pm 3.9 (–1.7)	87.7 \pm 1.3 (–20.9)	86 \pm 7.7 (–25.2)
					3g	Lit	100	45.5 \pm 4.7 (16.4)	100
50	50.9 \pm 3.9 (5.2)	71.6 \pm 7.2 (16.9)	64.3 \pm 10.1 (11.3)	67.7 \pm 2.2 (1.4)					
200	46.6 \pm 7.9 (13.2)	64.6 \pm 11.9 (25.0)	45.7 \pm 2.8 (36.9)	62.8 \pm 6.2 (8.5)					
3h	Lit	100			100	47.4 \pm 5.5 (11.7)	79.4 \pm 2.3 (7.9)	88.9 \pm 8.9 (–22.6)	97.2 \pm 1.4 (–41.5)
4a	H	100	44.3 \pm 4.3 (18.6)		100	33.3 \pm 13 (37.9)	48.3 \pm 6 (43.9)	61.6 \pm 1.6 (15)	53.3 \pm 9.2 (22.4)
					50	39.8 \pm 5.2 (25.9)	64.3 \pm 5.7 (25.4)	71.5 \pm 2.6 (1.4)	65.8 \pm 1.1 (4.2)
					200	49.8 \pm 7.4 (7.3)	86.7 \pm 10 (–0.6)	80.5 \pm 5.1 (–11)	85.5 \pm 8.1 (–24.5)
4b	CH ₃	100	46.3 \pm 5.6 (14.9)		100	35 \pm 2.8 (34.8)	80 \pm 17.3 (7.2)	80 \pm 7.6 (–10.3)	68.3 \pm 8.8 (0.6)
					50	32.5 \pm 4.9 (39.5)	68 \pm 7.4 (21.1)	66.7 \pm 6.8 (8)	59.7 \pm 6 (13.1)
					200	57.3 \pm 6.4 (–6.7)	104.1 \pm 11.6 (–20.7)	69.6 \pm 7.3 (4)	84.2 \pm 6.2 (–22.5)
4c	Br	100	36 \pm 6* (33.8)	7 \pm 0.4* (14.6)	100	42.5 \pm 2.5 (20.9)	70 \pm 15 (18.8)	55 \pm 5 (24.1)	50 \pm 10 (27.2)
					50	37.9 \pm 5.4 (29.4)	72.9 \pm 4.9 (15.4)	62.6 \pm 8.2 (13.7)	58.4 \pm 8 (15)
					200	49.4 \pm 3.7 (8)	77.7 \pm 8.3 (9.8)	57.9 \pm 5.7 (20.1)	65.3 \pm 4.9 (4.9)
4d	Cl	100	51 \pm 3.4 (6.3)		100	51.6 \pm 13 (3.9)	70 \pm 15 (18.8)	60 \pm 10 (17.2)	52.5 \pm 2.5 (23.6)
					50	43 \pm 1.5 (19.9)	72.9 \pm 8.1 (15.4)	58.2 \pm 4.5 (19.7)	55.7 \pm 5.3 (18.9)
					200	24.3 \pm 6.3* (54.7)	53.6 \pm 9.5 (37.8)	48.9 \pm 8.7 (32.6)	51.4 \pm 12.4 (25.1)
Control			54.4 \pm 4.2	8.2 \pm 0.2		53.7 \pm 8.9	86.2 \pm 7.4	72.5 \pm 12.6	68.7 \pm 12.8
ASA		200	27 \pm 3.9** (50.4)						
Morphine		10		4.5 \pm 0.3** (45.1)					
Indomethacin		10			10	47 \pm 5 (12.4)	60.5 \pm 4.9 (29.8)	51.1 \pm 1.2 (29.5)	44 \pm 3.4 (35.9)

*p < 0.05; **p < 0.01, significant from control.

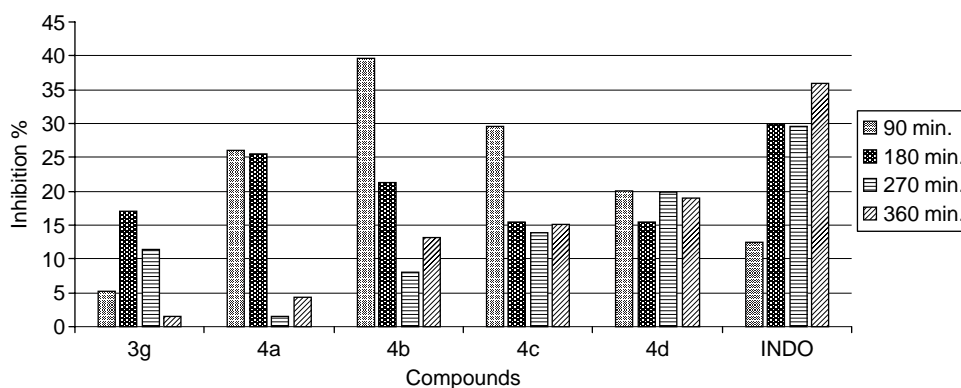


Figure 1. Effect of compounds against carrageenan-induced hind paw edema model at 50 mg/kg dose.

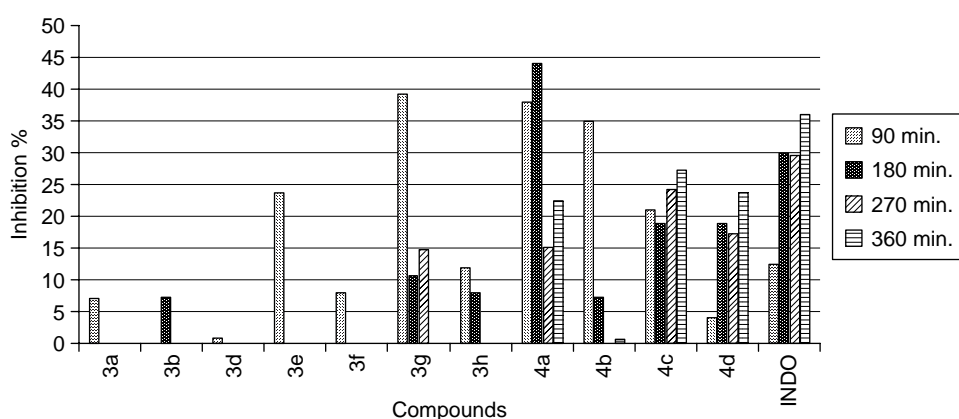


Figure 2. Effect of compounds against carrageenan-induced hind paw edema model at 100 mg/kg dose.

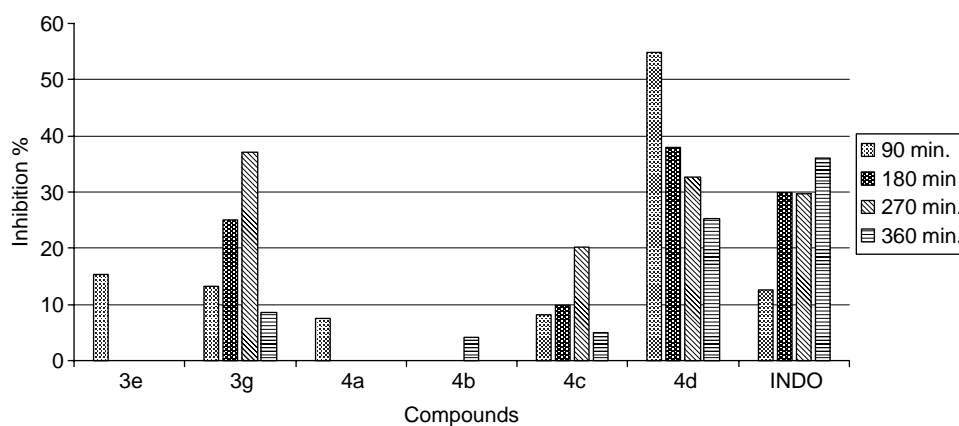


Figure 3. Effect of compounds against carrageenan-induced hind paw edema model at 200 mg/kg dose.

Analgesic activity. The analgesic activity of the compounds was studied by using both the acetic acid-induced writhing test [24] and hot plate test [25] in mice. The animals were first administered in 100 mg/kg (body weight) dose of the test drugs in two of the screening tests. Test compounds which possessed more than 20% effect at 100 mg/kg dose were evaluated for analgesic activity (Table I, Figure 4). Among the six derivatives investigated,

two compounds (3e, 4c) showed high activity. According to this, the compound 3e possessing 2,5-difluorobenzoyl in 6 position and 4-bromobenzoyl methyl in the third position of 2(3H)-benzoxazolone ring is the most active derivative (41.7%). Most of the structure-activity relationships seen for the anti-inflammatory activity were not confirmed for the analgesic activity. The analgesic activity of reduced derivatives 4a, 4b, 4d were not significant when

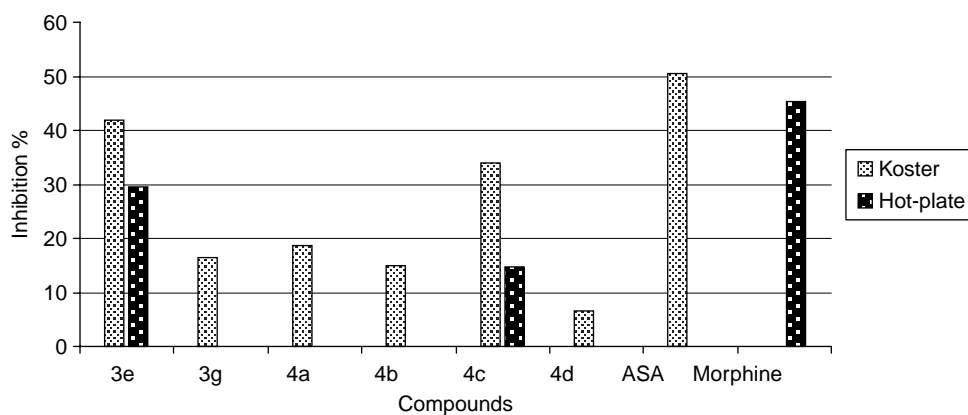


Figure 4. Analgesic effect of active compounds on acetic acid induced abdominal writhing and hot plate test.

compared to the activity obtained by Aspirin. The mouse abdominal constriction test can be used to detect both peripherally and centrally acting analgesics, whereas the hot-plate test has been demonstrated to be predictive of centrally acting analgesics. Therefore the analgesic activity of the two compounds was also determined by hot-plate test. Constant temperature hot-plate test result (Figure 2) was in good accordance with the Koster test for compound 3e. It showed moderately significant activity.

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